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AQA GCSE Chemistry 8462/8464 (9-1) Specification map

Tril	ogy, Chemistry only and Higher Tier	Unit 1	Unit 2	Unit 3	Unit 4	Unit 5	Unit 6	Unit 7	Unit 8	Unit 9	Unit 10 Unit 11
	omic structure and the periodic table						_			<u> </u>	
1.1	A simple model of the atom, symbols, relative atomic mass, electronic charge and isotopes										
1.2	The periodic table										
1.3	Properties of transition metals (chemistry only)										
2 Bo	nding, structure, and the properties of matter										
2.1	Chemical bonds, ionic, covalent and metallic			\bullet							
2.2	How bonding and structure are related to the properties of substances										
2.3	Structure and bonding of carbon			\bullet							
2.4	Bulk and surface properties of matter including nanoparticles (chemistry only)										
3 Qu	antitative chemistry										
3.1	Chemical measurements, conservation of mass and the quantitative interpretation of chemical equations				ullet						
3.2	Use of amount of substance in relation to masses of pure substances										
3.3	Yield and atom economy of chemical reactions (chemistry only)				•						
3.4	Using concentrations of solutions in mol/dm3 (chemistry only) (HT only)				•						
3.5	Use of amount of substance in relation to volumes of gases (chemistry only) (HT only)				ightarrow						
4 Ch	emical changes										
4.1	Reactivity of metals										
4.2	Reactions of acids										
4.3	Electrolysis										
5 En	ergy changes										
5.1	Exothermic and endothermic reactions										
5.2	Chemical cells and fuel cells (chemistry only)										



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	6 The rate and extent of chemical change			-		-	-	-	-		
6.1	Rate of reaction										
6.2	Reversible reactions and dynamic equilibrium										
7 Or	ganic chemistry										
7.1	Carbon compounds as fuels and feedstock										
7.2	Reactions of alkenes and alcohols (chemistry only)										
7.3	Synthetic and naturally occurring polymers (chemistry only)										
8 Ch	nemical analysis										
8.1	Purity, formulations and chromatography										
8.2	Identification of common gases										
8.3	Identification of ions by chemical and spectroscopic means (chemistry only)										
9 Ch	nemistry of the atmosphere										
9.1	The composition and evolution of the Earth's atmosphere										
9.2	Carbon dioxide and methane as greenhouse gases										
9.3	Common atmospheric pollutants and their sources										
10 U	lsing resources										
10.1	Using the Earth's resources and obtaining potable water										
10.2	Life cycle assessment and recycling										•
10.3	Using materials (chemistry only)										
10.4	The Haber process and the use of NPK fertilisers (chemistry only)										•